

**Formulas for Final, PHYS404, Fall 2014** v1.2

$$\Omega(N, N_{\uparrow}) = \frac{N!}{N_{\uparrow}!(N - N_{\uparrow})!}$$

$$\Omega(N, q) = \frac{(q + N - 1)!}{q!(N - 1)!}$$

$$\Omega(U, V, N) = f(N)V^N U^{fN/2}$$

$U = (f/2) Nk_B T$ ; determining  $f$ : 3 for atoms,  
5 for diatomic molecules at low  $T$ , 7 at higher  $T$ ,  
2 for each direction of a [n Einstein] oscillator  
Dimer rotational energy =  $j(j+1)\epsilon$

Ideal gas law  $pV = Nk_B T = nRT$   
van der Waals  $pV = Nk_B T / (V - Nb) - aN^2/V^2$

Entropy  $S = k_B \ln \Omega = \int dQ_{rev}/T$   $1/T = (\partial S / \partial U)_{V, N}$   
Ideal gas  $S = Nk_B [(V/(N\lambda_T^3)) + 5/2]$ ;  $\lambda_T = h/\sqrt{2\pi m k_B T}$   
Paramag  $S = Nk_B [\ln(2 \cosh x) - x \tanh x]$ ;  $x = \mu B / k_B T$   
 $U = -N\mu B \tanh(\beta\mu B)$ ;  $M = N\mu \tanh(\beta\mu B)$

$$\Delta U = Q + W_{on} = Q - W_{by} \quad \Delta S_{tot} \geq 0$$

Constants:  $k_B \approx 10^{-4}$  eV/K =  $1.38 \times 10^{-23}$  J/K  
 $C_V$  of 1 gm of water (ice) is 1 cal/K ( $\sim 1/2$  cal/K)  
 $R \sim 8.3$  J/K  $N_A = 6.02 \times 10^{23}$   
Stirling:  $\ln n! \approx n \ln n - n$

Expansions in  $\epsilon \ll 1$ :  
 $\ln(1 \pm \epsilon) \approx \pm \epsilon - \epsilon^2/2$ ,  $\exp(\pm \epsilon) \approx 1 \pm \epsilon + \epsilon^2/2!$   
 $\Gamma(n+1) = n!$  (integer  $n$ );  $\Gamma(1/2) = \sqrt{\pi}$ ;  $\Gamma(x+1) = x\Gamma(x)$

Change in internal energy, change in temperature,  
heat, work, during "simple" processes:  
isobaric ( $\Delta p = 0$ ), isochoric ( $\Delta V = 0 = W$ ), isothermal  
( $\Delta U = \Delta T = 0$ ), adiabatic ( $Q = 0$ ).  
Along an isobar,  $W_{by} = p(V_f - V_i)$ ; along an isotherm  
 $W_{by} = Nk_B T \ln(V_f/V_i)$   
Along an adiabat  $pV^\gamma$  is constant, as is (using the  
ideal gas law)  $TV^{\gamma-1}$  Note  $\gamma = (f+2)/f$

Helmholtz and Gibbs free energies  $F(T, V, [N]) = U - TS + \mu N$   
 $G(T, p, [N]) = U + pV - TS + \mu N$   
 $U(V, S, [N])$   $H(p, S, [N])$   $\Phi = U - TS - \mu N$

Thermo. identities:  $dU = TdS - p dV + \mu dN$   
 $dF = -S dT - p dV + \mu dN$   
 $dG = -S dT + V dp + \mu dN$ , etc.

Maxwell relations: 2<sup>nd</sup> deriv of thermo functions do  
not depend on order of derivs

$$\Delta S_{mix} = -Nk_B [x \ln x + (1-x) \ln(1-x)]$$

Expansion  $\beta = V^{-1} \partial V / \partial T|_p$   $\kappa_T = -V^{-1} \partial V / \partial p|_T$   
 $S = \int C/T dT$   $F = -k_B T \ln Z$   $U = -\partial \ln Z / \partial \beta$   
 $Z_N = Z_1^N / N!$  for indistinguishable  $\mu = (\partial F / \partial N)_{T, V}$

Nonrelativistic gas  $Z_1 = (V/\lambda_T^3) Z_{int}$ ;  $\lambda_T = h/(2\pi m k_B T)^{1/2}$

$$Z = \sum_i g_i e^{-\beta \epsilon_i} \quad \mathfrak{Z} = \sum_i g_i e^{-\beta(\epsilon_i - \mu N_i)} = \sum_N Z^N Z_N$$

[NB  $\epsilon_i$  means  $\epsilon_i$ ]

$Z_{rot} \approx k_B T / 2\epsilon$  for  $k_B T \gg \epsilon$ , dimer of identical atoms

$$\text{Maxwell 3D} \quad D(v) = \left( \frac{m}{2\pi k_B T} \right)^{3/2} 4\pi v^2 \exp\left( -\frac{mv^2}{2k_B T} \right)$$

Ideal gas  $\Phi = -k_B T \ln \mathfrak{Z} = -pV$   
 $d\Phi = -S dT - p dV - N d\mu$

$$\mathfrak{Z}_{FD/BE} = (1 \pm \exp[-\beta(\epsilon - \mu)])^{\pm 1} \quad \beta = 1/k_B T$$

$$\bar{n}_{FD/BE}(\epsilon; T, \mu) = \frac{1}{\exp[\beta(\epsilon - \mu)] \pm 1} = \frac{1}{z^{-1} \exp(\beta\epsilon) \pm 1}$$

$$\epsilon \propto n^s: \epsilon = \frac{h^2}{8mL^2} (n_x^2 + n_y^2 + n_z^2) = \frac{h^2 n^2}{8mL^2} \quad \text{or} \quad \epsilon = \frac{hcn}{2L}$$

$$E_F = (\hbar^2/2m)(3\pi^2 N_{el}/V)^{2/3} \quad \omega_D = (6\pi^2 N_{at}/V)^{1/3}$$

$$\mathcal{G}(\epsilon) \propto \epsilon^{(D/s)-1} / \epsilon_0^{(D/s)}$$

$$\int_0^{\epsilon_F} H(\epsilon) \bar{n}_{FD}(\epsilon; T) d\epsilon = \int_0^{\epsilon_F} H(\epsilon) d\epsilon + \frac{\pi^2}{6} (k_B T)^2 \left[ \frac{H'(\epsilon_F) - H(\epsilon_F) \mathcal{G}'(\epsilon_F)}{\mathcal{G}(\epsilon_F)} \right] \quad [\text{e.g.: } H = \mathcal{G} h]$$

$$\rightarrow +(\pi^2/6)(k_B T)^2 \mathcal{G}(\epsilon_F) h'(\epsilon_F)$$

$$\text{Photons } u(\epsilon) = [8\pi\epsilon^3/(hc)^3] \bar{n}_{BE}(\epsilon; T, 0) \quad \epsilon = hf$$

$$U/V = (8\pi^5/15)[(k_B T)^4/(hc)^3]$$

$$\text{Wien: } \lambda_{max} T = 0.0029 \text{ m K; } v_{max} \neq c/\lambda_{max}$$

Power from perfect blackbody radiator:  $\sigma \times \text{area} \times T^4$

$$\text{Debye } T_D = (\hbar c_s/k_B)(6\pi^2 N/V)^{1/3}$$

$$U = \frac{9Nk_B T^4}{T_D^3} \int_0^{T_D/T} \frac{x^3}{e^x - 1} dx$$

$$\text{BEC } \Gamma(\nu) Li_\nu(z) = \int_0^\infty \frac{x^{\nu-1} dx}{z^{-1} e^x - 1} \quad N = (V/\lambda_T^3) Li_{3/2}(z)$$

$$U = (3/2)(V/\lambda_T^3) k_B T Li_{5/2}(z) \quad N/V = 2.612/(\lambda_T(T=T_c))^3$$

$$E_{Ising} = -J \sum_{\langle i, j \rangle} s_i s_j, \quad s_i = \pm 1 \quad \text{MeanField } \bar{s} = \tanh(\beta q J \bar{s})$$