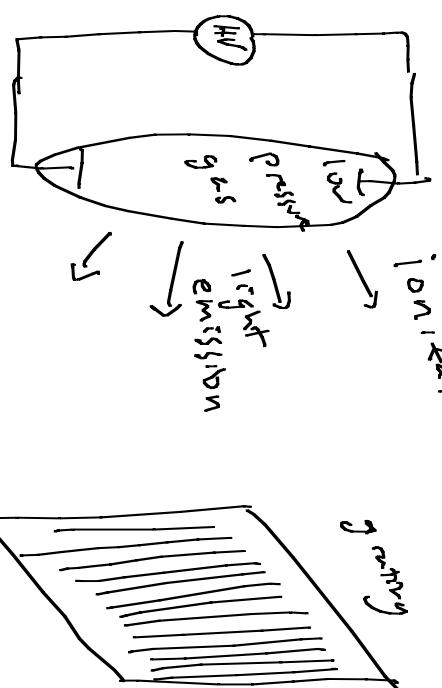
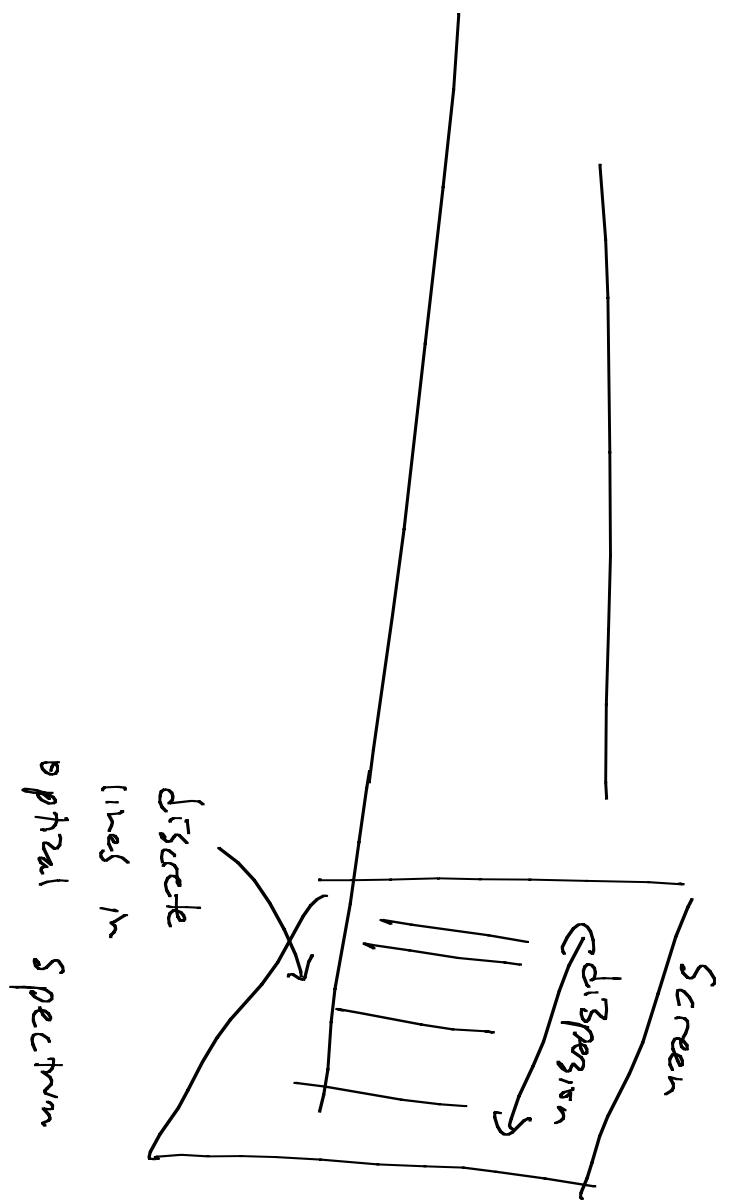


Atomic discharge



Spectroscopy



Bohr model

"Quantized" angular momentum

$$\textcircled{1} \quad mvr = n\hbar$$

Force balance

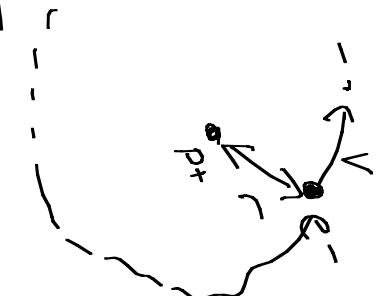
$$\textcircled{2} \quad \frac{mv^2}{r} = \frac{e^2}{r^2} \quad \Rightarrow \quad v = \sqrt{\frac{e^2}{mr}} \quad (*)$$

[Centripetal = Coulomb]

Subst. (*) into \textcircled{1},

$$\sqrt{\frac{e^2 m r}{r}} = n\hbar \quad \Rightarrow \quad r = \frac{n^2 \hbar^2}{e^2 m}$$

$$\begin{aligned} \text{Total Energy} &= \text{Kinetic} + \text{potential} = \frac{1}{2}mv^2 - \frac{e^2}{r} \\ \text{But, } (*) &\text{ gives Energy} = \frac{1}{2}mv^2 - \frac{e^2}{r} = -\frac{e^2}{2r} \end{aligned}$$



$$S_0 / \text{Energy} = -\frac{e^2}{2} \frac{e^{2m}}{h^2 c^2} = \left(\frac{e^2}{h c} \right)^2 \frac{mc^2}{2n^2}$$

$$\alpha = \frac{e^2}{h c} \quad \text{"fine structure constant"} \sim \frac{1}{137}$$

$$mc^2 = E_0 \quad \text{"rest mass"} \quad 511 \text{ keV}$$

$$\text{Energy} = -\alpha^2 \frac{E_0}{2n^2}$$

"Radiative transitions"



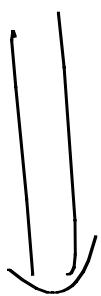
$$\Delta E = -\frac{R}{n'^2} - \left(-\frac{R}{n}\right) = R \left(\frac{1}{n^2} - \frac{1}{n'^2} \right)$$

Rydberg (1888)
113.6 eV

Fine Structure: electron magnetic moment (spin)

relativity:

$$\omega_e \downarrow$$



$$+ze$$

nucleus at rest

$$\omega_e \rightarrow$$

electron at rest

magnetic field at
electron due to charge
current from moving nucleus

$$\vec{B} = \frac{\mu_0}{4\pi} \frac{ze(-\vec{v} \times \vec{r})}{r^3} = -\frac{1}{4\pi\epsilon_0 c^2 m r^3} \vec{v}$$

$$(\vec{L} = m\vec{v} \times \vec{r})$$

"spin-orbit"

$$\text{Energy} = -\vec{m} \cdot \vec{B} = -\left(g \frac{\mu_B}{\hbar} \vec{s}\right) \cdot \left(\frac{1}{4\pi\epsilon_0 c^2 m r^3} \vec{v} \times \vec{L}\right) = \frac{1}{4\pi\epsilon_0 m c^2 r^3} \vec{s} \cdot \vec{L}$$

Since \vec{S} and \vec{L} are in (near-unity) units of \hbar , we have

$$\langle E_{\text{Spin-orbit}} \rangle \sim \frac{e^2}{m^2 c^2} \frac{\hbar^2}{r^3} \quad \text{and} \quad \langle r^3 \rangle \sim \left(\frac{\hbar^2}{e^2 m} \right)^3 \quad \text{so}$$

$$\langle E_{\text{Spin-orbit}} \rangle \sim \frac{e^8 m c^2}{\hbar^4 c^4} = \left(\frac{e^2}{\hbar c} \right)^4 m c^2 \rightarrow \text{factor of } \alpha^2 \text{ smaller than electronic state splitting!}$$

for fixed n , there are $2n-1$ possible values of L_z ($\vec{L} \cdot \hat{z}$) \rightarrow spin-orbit splitting of multiple states w/ same n .