

# Schrodinger Eq as a wave equation

If no forces, no potential energy:

$$i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2}, \quad \hbar \approx 10^{-34} \text{ J}\cdot\text{s}$$

Guess:  $\Psi(x,t) = Ae^{i(kx-\omega t)}$

Then  $i\hbar (-i\omega) Ae^{i(kx-\omega t)} = -\frac{\hbar^2}{2m} (ik)^2 Ae^{i(kx-\omega t)}$

$$\hbar\omega = \frac{\hbar^2 k^2}{2m}$$

$$\omega = \frac{\hbar k^2}{2m}$$

Because of (i) in the schrodinger Eq.,  $\Psi(x,t)$  must be complex.

∴ A general soln:  $\Psi(x,t) = \frac{1}{\sqrt{2\pi}} \int_{-\infty}^{\infty} dk A(k) e^{i(kx - \frac{\hbar k^2}{2m} t)}$

This system has dispersion. Group velocity,  $\frac{\partial \omega}{\partial k} = \frac{\hbar k}{m}$   
A gaussian pulse spreads out. Phase velocity =  $\frac{\omega}{k} = \frac{\hbar k}{2m}$

If forces are present, let the potential energy be  $V(x)$  ( $F = -\frac{dV}{dx}$ ). Then the schrodinger Eq. is

$$i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + V(x)\Psi(x)$$

Group velocity interpretation: classical Velocity =  $\frac{\text{momentum}}{\text{mass}} = \frac{p}{m}$   
∴  $\hbar k \rightarrow \text{momentum}$

To solve the Schrodinger Eq we guess solutions like

$$\Psi(x,t) = \psi(x)e^{-i\omega t}$$

interpretation:  $\omega = \frac{\text{Energy}}{\hbar}$

Any solution of this type is called an "energy eigenstate".

$$= \psi(x)e^{-\frac{iE}{\hbar}t}$$

$$\hbar \approx 10^{-34} \text{ J}\cdot\text{s}$$

Then ~~(i\hbar)~~  $\frac{\partial}{\partial t}$

$$\text{Then } (i\hbar) \left( \frac{-iE}{\hbar} \right) \psi(x) = -\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2}$$

$$\boxed{\psi''(x) = -\frac{2mE}{\hbar^2} \psi(x)}$$

For positive E, Curvature of  $\psi$  is opposite sign from  $\psi$ .

$$\text{or } \psi''(x) + \frac{2mE}{\hbar^2} \psi(x) = 0 \quad \Leftarrow$$

$$\text{or } -\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} = E\psi \quad \Leftarrow \text{Time Independent Schrodinger Eq.}$$

"Normal Mode" in classical physics (Phys 273)

$\Rightarrow$  "Energy Eigenstate" in Quantum Mechanics

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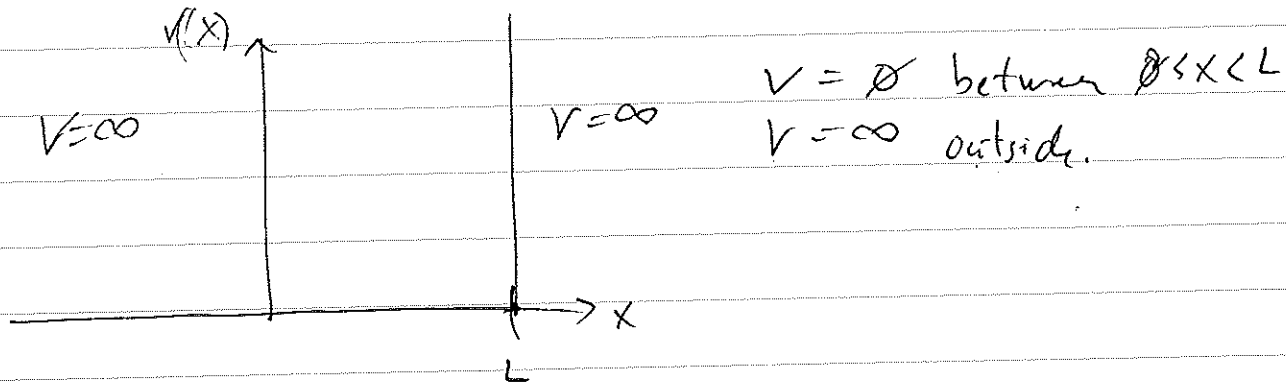
More quantum interpretation. The wavefunction squared is the probability to find the particle at position  $x$ .  $P(x) = |\psi(x)|^2$

Total probability must be 1:

$$\Rightarrow \int_{-\infty}^{\infty} |\psi(x)|^2 dx \equiv 1.$$

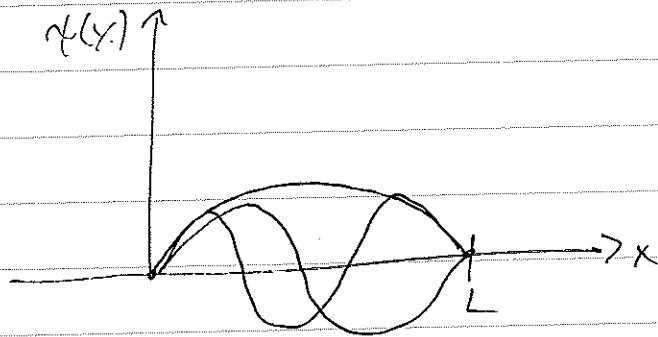
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Simple case: Particle in a box,



It turns out the  $\psi$  must be zero where  $V = \infty$ , and that  $\psi$  must be continuous. This is very ~~precisely~~ similar to the case where a rope is fixed btw two walls at  $x = 0$  and  $x = L$ .

Need  $\psi(x) = A \sin\left(\frac{n\pi x}{L}\right)$ ,  $n = \text{integer}$



$$k_n = \frac{n\pi}{L}$$

$$\omega_n = \frac{\hbar k_n^2}{2m} = \frac{\hbar n^2 \pi^2}{2mL^2}$$

$$E_n = \hbar \omega_n = \frac{\hbar^2 n^2 \pi^2}{2mL^2}$$

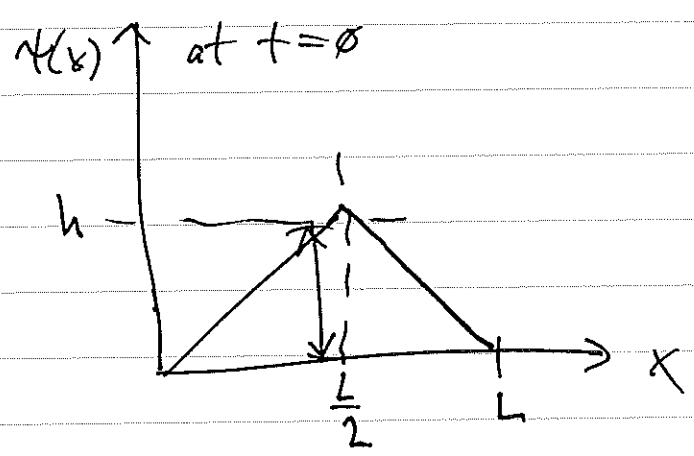
$$\psi(x,t) = A \sin\left(\frac{n\pi x}{L}\right) e^{i\omega_n t} = A \sin\left(\frac{n\pi x}{L}\right) e^{i\left(\frac{\hbar n^2 \pi^2}{2mL^2} t\right)}$$

$$\text{General Solution} = \psi(x,t) = \sum_{n=1}^{\infty} \frac{a_n}{a_n} A_n \sin\left(\frac{n\pi x}{L}\right) e^{i\omega_n t}$$

Suppose we are told what  $\psi(x,t)$  looks like at  $t=0$ . How do we calculate the correct expansion coefficients  $\{a_n\}$ ? Answer: Fourier's Trick.

$$a_n = \frac{2}{L} \int_0^L \psi(x, t=0) \sin\left(\frac{n\pi x}{L}\right) dx$$

Example: Triangle function at  $t=0$ :



Then  $a_n = \frac{8h}{(n\pi)^2} (-1)^{(n-1)/2}$ ,  $n = \text{odd only}$ .

$$\psi(x,t) = \sum_{n=1}^{\infty} \frac{8h}{(n\pi)^2} (-1)^{(n-1)/2} \sin\left(\frac{n\pi x}{L}\right) e^{-i\left(\frac{n^2\pi^2}{L^2}t\right)}$$

Need to choose the constant  $h$  so that  $\int \psi^2 dx = 1$ .

The primary distinction with the rope system is that  $\omega_n$  for the rope

$$\omega_n = \sqrt{\frac{T}{\mu}} \frac{n\pi}{L}$$

In QM the expansion coefficients have a profound physical interpretation: They are proportional to the probability of observing energy values:

$|a_n|^2 \sim$  probability to observe energy  $E_n$  when for the particle in a box

$$E_n = \frac{n^2 \pi^2 \hbar^2}{2mL^2}$$

This means  $\sum_{n=1}^{\infty} |a_n|^2 = 1$

~~Ex:  $\psi(x,t) = \dots$~~

In general, let  $\{\alpha_n(x)\}$  be the set of energy eigenfunctions ( $\sim \sqrt{\frac{2}{L}} \sin(\frac{n\pi x}{L})$  in this case).

$$\text{Then } \Psi(x,t) = a_1 \alpha_1(x) e^{-i\omega_1 t} + a_2 \alpha_2(x) e^{-i\omega_2 t} + a_3 \alpha_3(x) e^{-i\omega_3 t} + \dots$$

Then  $|a_1|^2 =$  Probability to observe energy  $= \frac{E_1}{\hbar\omega_1} = \hbar\omega_1$   
 $|a_2|^2 = \dots = E_2 = \hbar\omega_2$

If  $V(x) = \text{constant} = V_0$ , then again

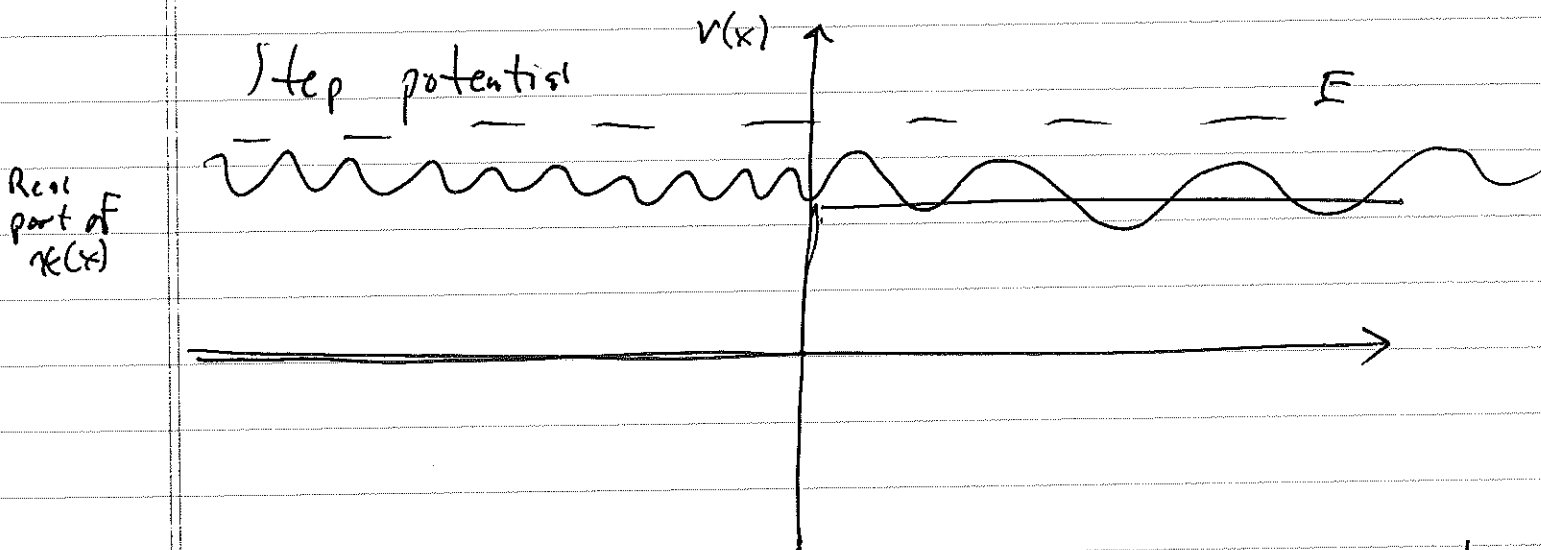
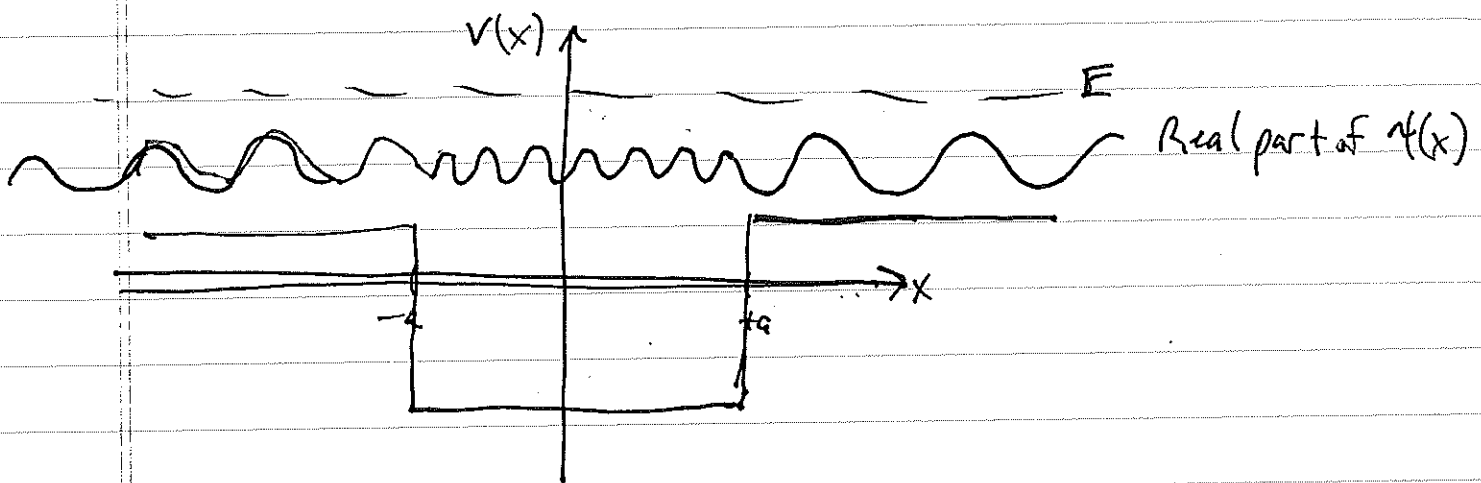
(6)

$$\psi(x,t) = A e^{i(kx - \omega t)}$$

now  $k$  must be  $k = \sqrt{\frac{2m(E - V_0)}{\hbar^2}}$

where  $E$  is the particle energy,  $E = \frac{(\hbar k)^2}{2m} + V_0$

finite square well:



Calculate Transmission & Reflection coefficients for different types of potential steps.

Why "Quantum" Mechanics? What is quantized?

For free particles, nothing is quantized. For free particles ( $V = \phi$ ), "quantum" mechanics is a misnomer. The name comes from the case of bound particles, like the particle in a box.

The energy is quantized:  $E_n = \hbar \omega_n = \frac{\hbar^2 n^2 \pi^2}{2mL^2}$   
(discrete)

For free particles,  $E = \frac{\hbar^2 k^2}{2m}$ , all  $k$  are allowed, because there are no boundary conditions.

~~For bound particles the Schrodinger Eq. says~~

~~$$\psi''(x) = -\frac{2m(E - V(x))}{\hbar^2}$$~~