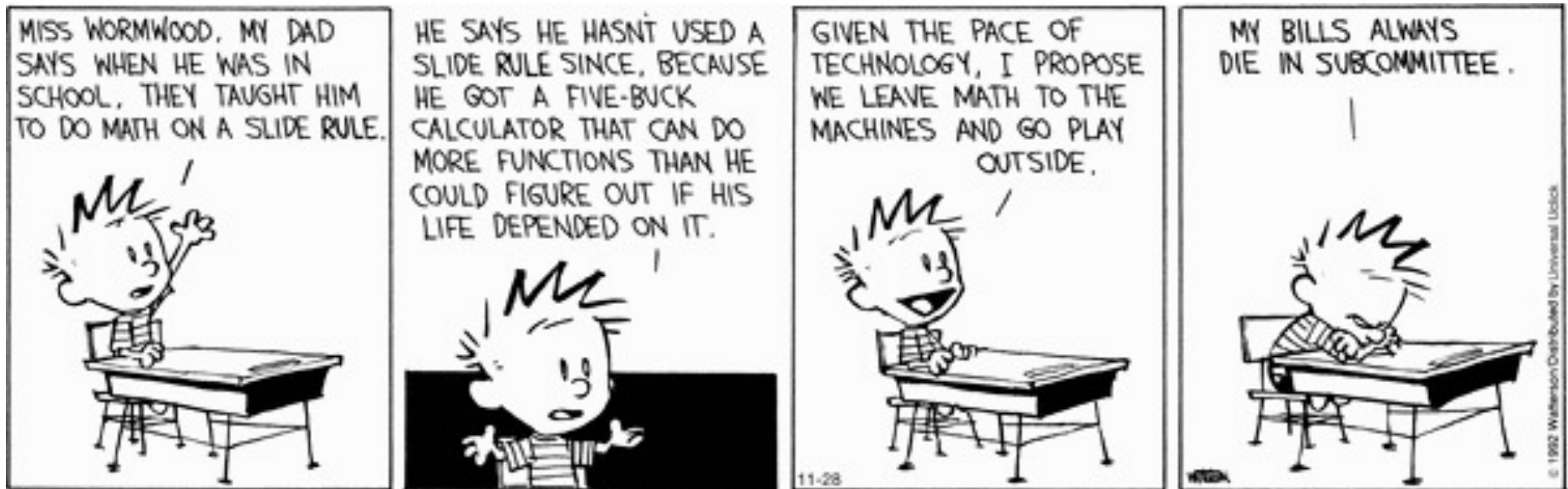


■ Theme Music: Doris Day

*Que Sera, Sera*

■ Cartoon: Bill Watterson

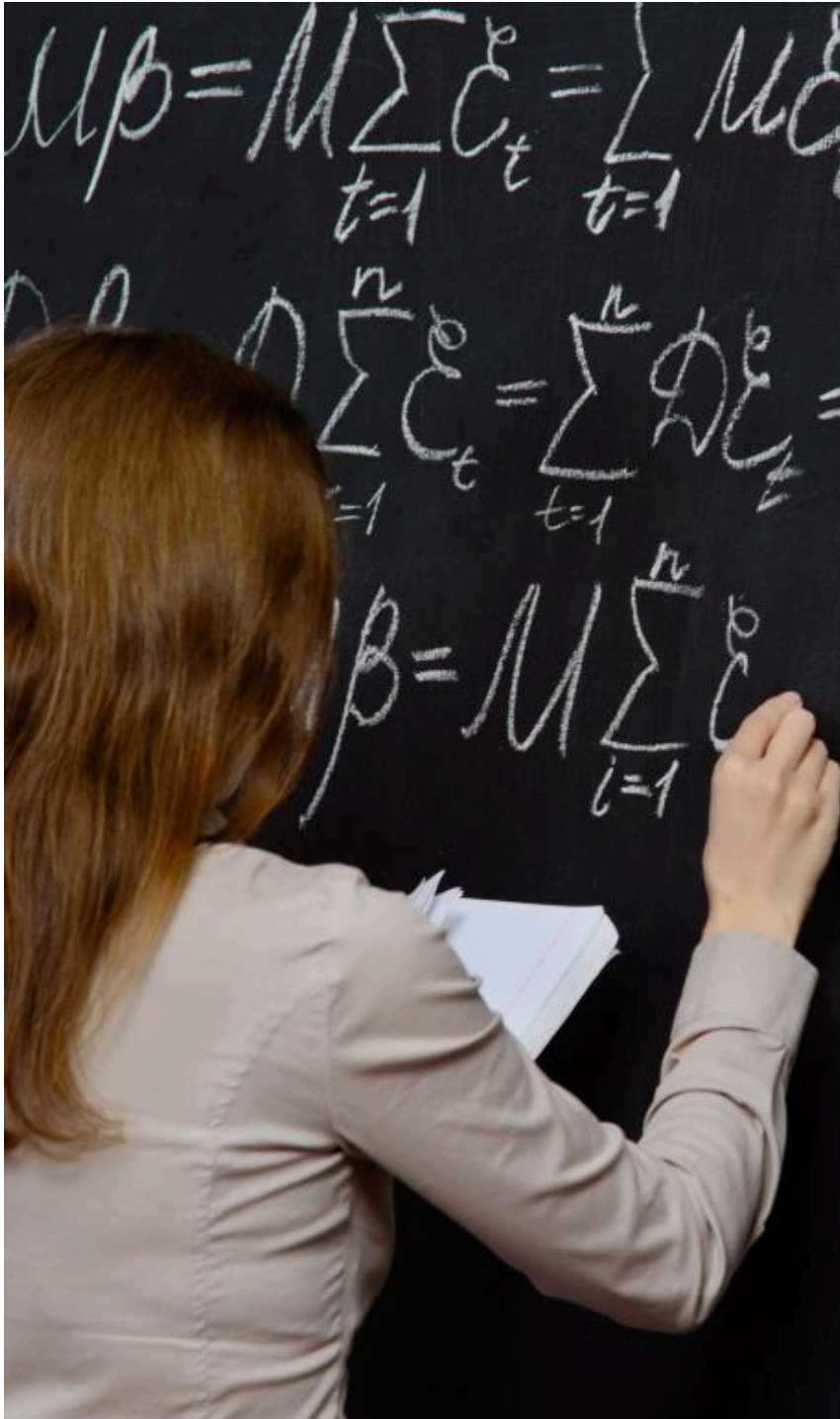
*Calvin & Hobbes*



# The Equation of the Day

Entropy  
(Information  
definition)

$$S = k_B \ln W$$



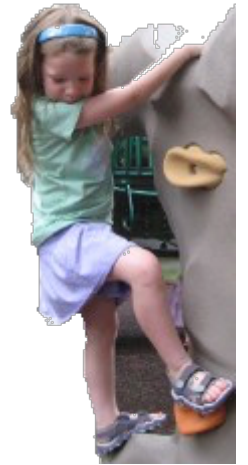
# Propose a formula



- Suppose I have a block of matter with  $N$  two-state “Degrees of Freedom” (bins in which to place energy that can only hold 1 energy packet).
- I have  $M$  packets of thermal energy. How many ways are there to distribute  $M$  packets?

- Hint: “ $N$  choose  $M$ ” = 
$$C_{N,M} = \frac{N!}{M!(N-M)!}$$

# Foothold ideas: Exponents and logarithms



- Power law:  $f(x) = x^2$   $g(x) = Ax^7$   
a variable raised to a fixed power.

- Exponential:  $f(x) = e^x$   $g(N) = 2^N$   $h(z) = 10^z$   
a fixed constant raised to a variable power.

- Logarithm: the inverse  
of the exponential.

$$x = e^{\ln(x)} \quad x = \ln(e^x)$$

$$y = 10^{\log(y)} \quad y = \log(10^y)$$

$$\log(2) = 0.3010$$

$$\log(e) = 0.4343$$

$$2^N = (10^{0.3010})^N \approx 10^{0.3N}$$

$$e^x = (10^{0.4343})^x \approx 10^{0.4x}$$

$$2^N = B$$

$$N \log 2 = \log B \Rightarrow N = \frac{\log B}{\log 2}$$

Logs convert multiplying to adding!