Consider the reaction $C + O_2 \rightarrow CO_2$



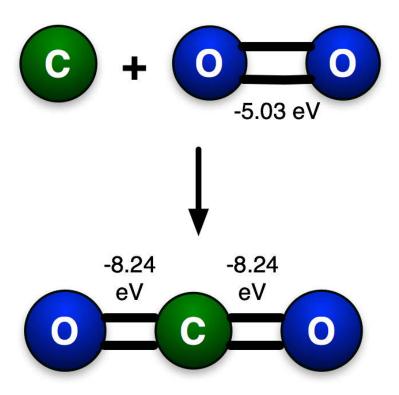
What is the **magnitude** of the change in the chemical energy for this reaction (per reaction)?

A. 8.24 eV

B. 3.21 eV

C. 11.45 eV

D. 16.48 eV



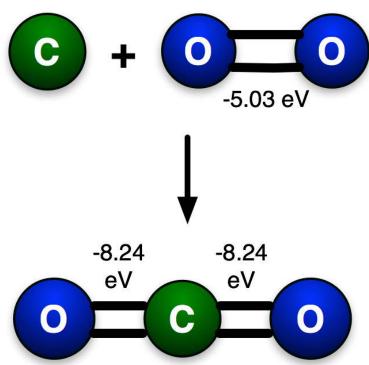
Consider the reaction



 $C + O_2 \rightarrow CO_2$ What is the **sign** of the

change in the chemical energy for this reaction (per reaction)?

- A. Positive
- B. Negative
- C. Cannot be determined



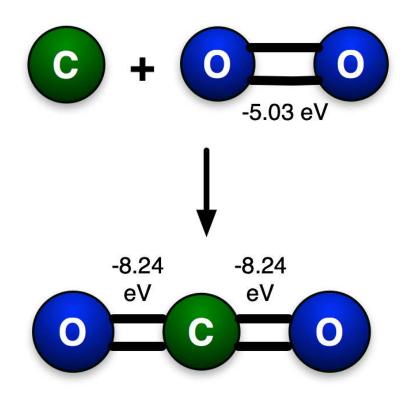
Consider the reaction

$$C + O_2 \rightarrow CO_2$$



If a mixture of carbon and oxygen gas undergoes the reaction at the right, what is the **sign** of the change in the **thermal** energy?

- A. Positive
- B. Negative
- C. Cannot be determined



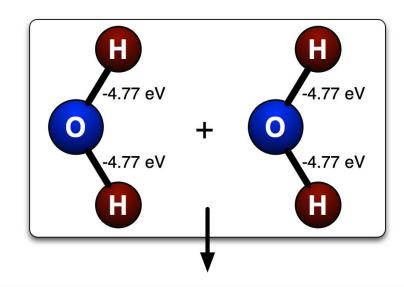
Physic Physic Physic your everyday experiences?

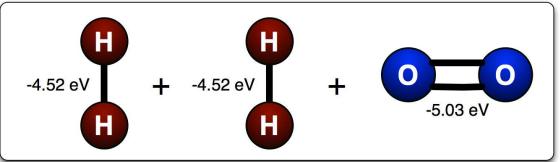
Consider the reaction $2H_2O \rightarrow 2H_2 + O_2$



What is the **sign** of the change in the chemical energy for this reaction (per reaction)?

- A. Positive
- B. Negative
- C. Cannot be determined





Does this make sense with your everyday experiences?

Consider the reaction $2H_2O \rightarrow 2H_2 + O_2$



What is the magnitude of the change in the chemical energy for this reaction (per reaction)?

2/1/16

