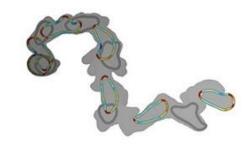
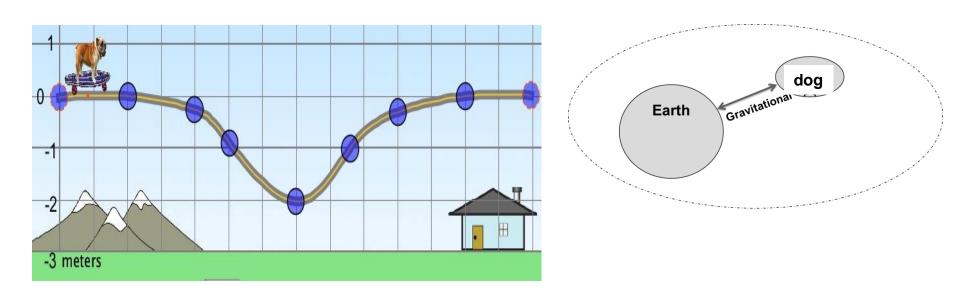
# Physics 131- Fundamentals of Physics for Biologists I



Forces from Potential Energy Binding Energy

### Conservation of Total Mechanical Energy



### Forces from Potential Energy (PE)

> For conservative forces, PE can be defined by

$$\vec{F} \cdot \Delta \vec{r} = -\Delta U$$

> If you know *U*, the force can be gotten from it via

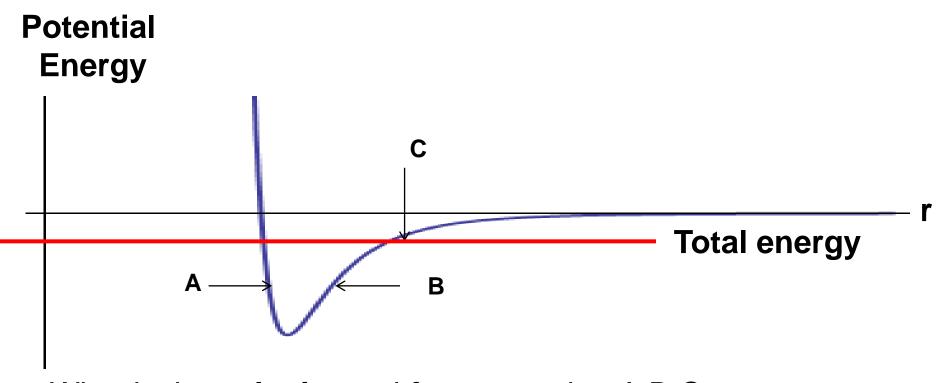
$$F_{\parallel}^{type} = -\frac{\Delta U_{type}}{\Delta r} = -\frac{dU_{type}}{dr}$$

In more than 1D need to use the gradient

$$\vec{F}^{type} = \cdot - \vec{\nabla} U_{type}$$



> The force always points down the PE hill.

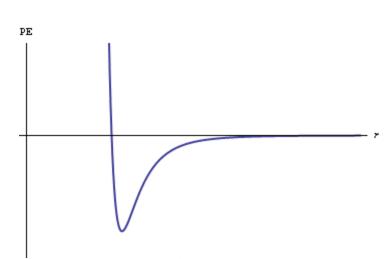


What is the **velocity** and **force** at point A,B,C Consider both magnitude and direction!

Draw the vectors on the whiteboard

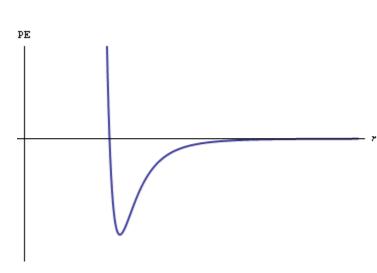
Whiteboard, TA & LA

You know that two atoms that are far apart are barely interacting. How is this represented visually in the PE diagram?



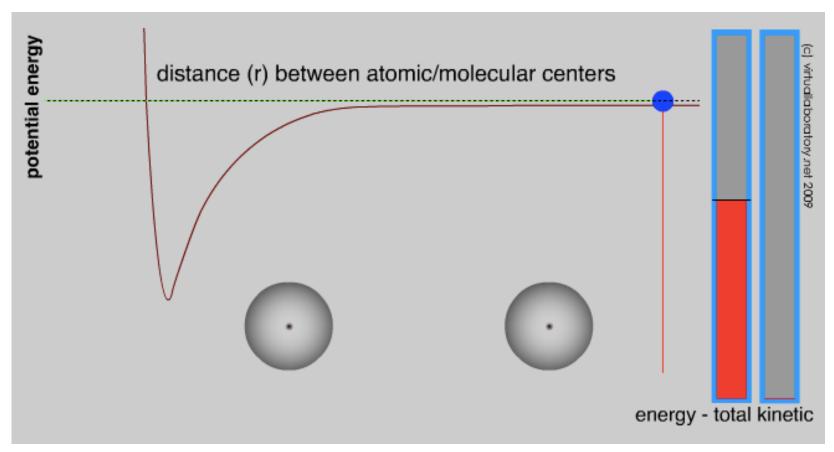
- The potential energy approaches zero as r gets large.
- 2. The PE curve is close to horizontal as r gets large.
- 3. The PE curve is close to vertical as r gets small.
- 4. The potential energy has a minimum.
- 5. More than one of these
- 6. The PE diagram doesn't demonstrate this information
- None of these

These two atoms can exist in a stable bound state. How is this represented visually in the PE diagram?



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- 2. The PE curve is close to horizontal as r gets large.
- 3. The PE curve is close to vertical as r gets small.
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#### Molecular forces



http://besocratic.colorado.edu/CLUE-Chemistry/activities/LondonDispersionForce/1.2-interactions-0.html

## Energies between charge clusters

- Atoms and molecules are made up of charges.
- The potential energy between two charges is

$$U_{12}^{elec} = \frac{k_C Q_1 Q_2}{r_{12}}$$
 No vectors!

The potential energy between many charges is

$$U_{12...N}^{elec} = \sum_{i < j=1}^{N} \frac{k_{C}Q_{i}Q_{j}}{r_{ij}}$$
 Just add up all pairs!

Physics 131