

CHAPTER 5

PROBLEM 5.1(a) $(m_1 + m_2)\vec{R} = m_1\vec{r}_1 + m_2\vec{r}_2 = m_1\vec{r}_1 + m_2(\vec{r}_1 - \vec{r}_2) = (m_1 + m_2)\vec{r}_1 - m_2\vec{r}_2 \Rightarrow$

$$\vec{r}_1 = \vec{R} + \frac{m_2}{m_1 + m_2} \vec{r} = \vec{R} + \frac{\mu}{m_1} \vec{r} \quad \checkmark$$

$$(m_1 + m_2)\vec{R} = m_1(\vec{r}_2 + \vec{r}) + m_2\vec{r}_2 = (m_1 + m_2)\vec{r}_2 + m_1\vec{r} \Rightarrow \vec{r}_2 = \vec{R} - \frac{m_1}{m_1 + m_2} \vec{r} = \vec{R} - \frac{\mu}{m_2} \vec{r} \quad \checkmark$$

Let $\vec{R} = (X, Y, Z)$, $\vec{r} = (x, y, z)$. $(\nabla_1)_x = \frac{\partial}{\partial x_1} = \frac{\partial X}{\partial x_1} \frac{\partial}{\partial X} + \frac{\partial x}{\partial x_1} \frac{\partial}{\partial x} = \left(\frac{m_1}{m_1 + m_2}\right) \frac{\partial}{\partial X} + (1) \frac{\partial}{\partial x} = \frac{\mu}{m_2} (\nabla_R)_x + (\nabla_r)_x$

So $\vec{\nabla}_1 = \frac{\mu}{m_2} \vec{\nabla}_R + \vec{\nabla}_r \quad \checkmark$

$$(\nabla_2)_x = \frac{\partial}{\partial x_2} = \frac{\partial X}{\partial x_2} \frac{\partial}{\partial X} + \frac{\partial x}{\partial x_2} \frac{\partial}{\partial x} = \left(\frac{m_2}{m_1 + m_2}\right) \frac{\partial}{\partial X} - (1) \frac{\partial}{\partial x} = \frac{\mu}{m_1} (\nabla_R)_x - (\nabla_r)_x \Rightarrow \vec{\nabla}_2 = \frac{\mu}{m_1} \vec{\nabla}_R - \vec{\nabla}_r \quad \checkmark$$

$$\begin{aligned} \text{(b) } \nabla_1^2 \psi &= \vec{\nabla}_1 \cdot (\vec{\nabla}_1 \psi) = \vec{\nabla}_1 \cdot \left[\frac{\mu}{m_2} \vec{\nabla}_R \psi + \vec{\nabla}_r \psi \right] = \frac{\mu}{m_2} \vec{\nabla}_R \cdot \left(\frac{\mu}{m_2} \vec{\nabla}_R \psi + \vec{\nabla}_r \psi \right) + \vec{\nabla}_r \cdot \left(\frac{\mu}{m_1} \vec{\nabla}_R \psi + \vec{\nabla}_r \psi \right) \\ &= \left(\frac{\mu}{m_2}\right)^2 \nabla_R^2 \psi + 2 \frac{\mu}{m_2} (\vec{\nabla}_r \cdot \vec{\nabla}_R) \psi + \nabla_r^2 \psi. \text{ Likewise } \nabla_2^2 \psi = \left(\frac{\mu}{m_1}\right)^2 \nabla_R^2 \psi - 2 \frac{\mu}{m_1} (\vec{\nabla}_r \cdot \vec{\nabla}_R) \psi + \nabla_r^2 \psi. \end{aligned}$$

$$\therefore H\psi = -\frac{\hbar^2}{2m_1} \nabla_1^2 \psi - \frac{\hbar^2}{2m_2} \nabla_2^2 \psi + V(\vec{r}_1, \vec{r}_2) \psi = -\frac{\hbar^2}{2} \left\{ \frac{\mu^2}{m_1 m_2} \nabla_R^2 + \frac{2\mu}{m_1 m_2} \vec{\nabla}_r \cdot \vec{\nabla}_R + \frac{1}{m_1} \nabla_r^2 + \frac{\mu^2}{m_2 m_1} \nabla_R^2 - \frac{2\mu}{m_2 m_1} \vec{\nabla}_r \cdot \vec{\nabla}_R + \frac{1}{m_2} \nabla_r^2 \right\} \psi$$

$$+ V(\vec{r}) \psi = -\frac{\hbar^2}{2} \left[\frac{\mu^2}{m_1 m_2} \left(\frac{1}{m_2} + \frac{1}{m_1} \right) \nabla_R^2 + \left(\frac{1}{m_1} + \frac{1}{m_2} \right) \nabla_r^2 \right] \psi + V(\vec{r}) \psi = E\psi. \text{ But } \left(\frac{1}{m_1} + \frac{1}{m_2} \right) = \frac{m_1 + m_2}{m_1 m_2} = \frac{1}{\mu}, \text{ so}$$

$$\frac{\mu^2}{m_1 m_2} \left(\frac{1}{m_2} + \frac{1}{m_1} \right) = \frac{\mu}{m_1 m_2} = \frac{m_1 m_2}{m_1 m_2 (m_1 + m_2)} = \frac{1}{m_1 + m_2}. \therefore -\frac{\hbar^2}{2(m_1 + m_2)} \nabla_R^2 \psi - \frac{\hbar^2}{2\mu} \nabla_r^2 \psi + V(\vec{r}) \psi = E\psi. \quad \checkmark$$

(c) Put in $\psi = \psi_r(\vec{r}) \psi_R(\vec{R})$, and divide by $\psi_r \psi_R$:

$$\left[-\frac{\hbar^2}{2(m_1 + m_2)} \frac{1}{\psi_R} \nabla_R^2 \psi_R \right] + \left[-\frac{\hbar^2}{2\mu} \frac{1}{\psi_r} \nabla_r^2 \psi_r + V(\vec{r}) \right] = E. \text{ First term depends only on } \vec{R}, \text{ second only on } \vec{r},$$

So each must be a constant — call them E_R and E_r , respectively. Then:

$$\boxed{-\frac{\hbar^2}{2(m_1 + m_2)} \nabla^2 \psi_R = E_R \psi_R} ; \quad \boxed{-\frac{\hbar^2}{2\mu} \nabla^2 \psi_r + V(\vec{r}) \psi_r = E_r \psi_r}, \text{ with } \boxed{E_R + E_r = E}.$$

PROBLEM 5.2 (a) From [4.77], E_1 is proportional to mass, so $\frac{\Delta E_1}{E_1} = \frac{\Delta m}{\mu} = \frac{m - \mu}{\mu} = \frac{m(m + M)}{mM} - \frac{M}{M} = \frac{m}{M}$.

So the fractional error is the ratio of the electron mass to the proton mass: $\frac{9.109 \times 10^{-31} \text{ kg}}{1.673 \times 10^{-27} \text{ kg}} = 5.44 \times 10^{-4}$.

The percent error is $\boxed{0.054\%}$ (pretty small).

(b) From [4.94], R is proportional to m , so $\frac{\Delta(1/\lambda)}{(1/\lambda)} = \frac{\Delta R}{R} = \frac{\Delta \mu}{\mu} = -\frac{(1/\lambda^2) \Delta \lambda}{(1/\lambda)} = -\frac{\Delta \lambda}{\lambda}$.

So (in magnitude) $\frac{\Delta \lambda}{\lambda} = \frac{\Delta \mu}{\mu}$. $\mu = \frac{mM}{m + M}$, where m = electron mass, and M = nuclear mass.

$$\Delta \mu = \frac{m \Delta M}{M + 2m} - \frac{mM}{M + 2m} = \frac{m \Delta M}{(M + 2m)(M + 2m)} [2M + 2m - M - 2m] = \frac{m \Delta M}{(M + 2m)(M + 2m)} = \frac{m \Delta M}{M + 2m}$$

$$\frac{\Delta \lambda}{\lambda} = \frac{\Delta \mu}{\mu} = \frac{m}{M + 2m} \approx \frac{m}{2m_p}, \text{ so } \boxed{\Delta \lambda = \frac{m}{2m_p} \lambda_h}, \text{ where } \lambda_h \text{ is the hydrogen wavelength.}$$

$$\frac{1}{\lambda} = R \left(\frac{1}{4} - \frac{1}{9} \right) = \frac{5}{36} R \Rightarrow \lambda = \frac{36}{5R} = \frac{36}{5(1.097 \times 10^7)} \text{ m} = 6.563 \times 10^{-7} \text{ m}. \quad \Delta \lambda = \frac{9.109 \times 10^{-31}}{2(1.673 \times 10^{-27})} (6.563 \times 10^{-7}) \text{ m} = \boxed{1.79 \times 10^{-10} \text{ m}}$$

(c) $\mu = \frac{m m_\mu}{m + m_\mu} = \frac{m}{2}$, so the energy is half what it would be for hydrogen: $\frac{13.6 \text{ eV}}{2} = \boxed{6.8 \text{ eV}}$.

(d) $\mu = \frac{m m_\mu}{m + m_\mu}$; $R \propto \mu$, so R is changed by a factor $\frac{m_p m_\mu}{m_p + m_\mu} \cdot \frac{m_p + m_e}{m_p m_e} = \frac{m_\mu (m_p + m_e)}{m_e (m_p + m_\mu)}$, as compared with hydrogen. For hydrogen, $\frac{1}{\lambda} = R (1 - \frac{1}{4}) = \frac{3}{4} R \Rightarrow \lambda = \frac{4}{3R} = \frac{4}{3(1.097 \times 10^7)} \text{ m} = 1.215 \times 10^{-7} \text{ m}$, and $\lambda \propto \frac{1}{R}$, so for muonic hydrogen the Lyman-alpha line is at

$$\lambda = \frac{m_e (m_p + m_\mu)}{m_\mu (m_p + m_e)} (1.215 \times 10^{-7} \text{ m}) = \frac{1}{206.77} \frac{(1.673 \times 10^{-27} + 206.77 \times 9.109 \times 10^{-31})}{(1.673 \times 10^{-27} + 9.109 \times 10^{-31})} (1.215 \times 10^{-7} \text{ m}) = \boxed{6.54 \times 10^{-10} \text{ m}}$$

PROBLEM 5.3 (a) $1 = \int |\psi_\pm|^2 d^3r_1 d^3r_2 = |A|^2 \int [\psi_a(r_1)\psi_b(r_2) \pm \psi_b(r_1)\psi_a(r_2)]^* [\psi_a(r_1)\psi_b(r_2) \pm \psi_b(r_1)\psi_a(r_2)] d^3r_1 d^3r_2$
 $= |A|^2 \{ \int |\psi_a(r_1)|^2 d^3r_1 \int |\psi_b(r_2)|^2 d^3r_2 \pm \int \psi_a(r_1)^* \psi_b(r_1) d^3r_1 \int \psi_b(r_2)^* \psi_a(r_2) d^3r_2 \pm \int \psi_b(r_1)^* \psi_a(r_1) d^3r_1 \int \psi_a(r_2)^* \psi_b(r_2) d^3r_2 + \int |\psi_b(r_1)|^2 d^3r_1 \int |\psi_a(r_2)|^2 d^3r_2 \} = |A|^2 (1 \cdot 1 \pm 0 \cdot 0 \pm 0 \cdot 0 + 1 \cdot 1) = 2|A|^2 \Rightarrow \boxed{A = \frac{1}{\sqrt{2}}}$.

(b) $1 = |A|^2 \int [2\psi_a(r_1)\psi_a(r_2)]^* [2\psi_a(r_1)\psi_a(r_2)] d^3r_1 d^3r_2 = 4|A|^2 \int |\psi_a(r_1)|^2 d^3r_1 \int |\psi_a(r_2)|^2 d^3r_2 = 4|A|^2 \cdot \boxed{A = \frac{1}{2}}$.

PROBLEM 5.4 (a) $-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx_1^2} - \frac{\hbar^2}{2m} \frac{d^2\psi}{dx_2^2} = E\psi$ (for $0 \leq x_1, x_2 \leq a$ - otherwise $\psi = 0$).

$$\psi = \frac{\sqrt{2}}{a} [\sin(\frac{\pi x_1}{a}) \sin(\frac{2\pi x_2}{a}) - \sin(\frac{2\pi x_1}{a}) \sin(\frac{\pi x_2}{a})] \Rightarrow \frac{d^2\psi}{dx_1^2} = \frac{\sqrt{2}}{a} [-(\frac{\pi}{a})^2 \sin(\frac{\pi x_1}{a}) \sin(\frac{2\pi x_2}{a}) - (-\frac{2\pi}{a})^2 \sin(\frac{2\pi x_1}{a}) \sin(\frac{\pi x_2}{a})]$$

$$\frac{d^2\psi}{dx_2^2} = \frac{\sqrt{2}}{a} [-(\frac{2\pi}{a})^2 \sin(\frac{\pi x_1}{a}) \sin(\frac{2\pi x_2}{a}) - (-\frac{\pi}{a})^2 \sin(\frac{2\pi x_1}{a}) \sin(\frac{\pi x_2}{a})], \text{ so } (\frac{d^2\psi}{dx_1^2} + \frac{d^2\psi}{dx_2^2}) = -[(\frac{\pi}{a})^2 + (\frac{2\pi}{a})^2] \psi = -5 \frac{\pi^2}{a^2} \psi.$$

$$\therefore -\frac{\hbar^2}{2m} (\frac{d^2\psi}{dx_1^2} + \frac{d^2\psi}{dx_2^2}) = \frac{5\hbar^2}{2m} \frac{\pi^2}{a^2} \psi = E\psi, \text{ with } E = \frac{5\hbar^2 \pi^2}{2ma^2} = 5K. \checkmark$$

(b) Distinguishable: $\psi_{22} = \frac{2}{a} \sin(\frac{2\pi x_1}{a}) \sin(\frac{2\pi x_2}{a})$, with $E_{22} = 8K$ (nondegenerate)

$$\left. \begin{aligned} \psi_{13} &= \frac{2}{a} \sin(\frac{\pi x_1}{a}) \sin(\frac{3\pi x_2}{a}) \\ \psi_{31} &= \frac{2}{a} \sin(\frac{3\pi x_1}{a}) \sin(\frac{\pi x_2}{a}) \end{aligned} \right\} \text{ with } E_{13} = E_{31} = 10K \text{ (doubly degenerate)}$$

Identical Bosons: $\psi_{22} = \frac{2}{a} \sin(\frac{2\pi x_1}{a}) \sin(\frac{2\pi x_2}{a})$, $E_{22} = 8K$ (nondegenerate)

$$\psi_{13} = \frac{\sqrt{2}}{a} [\sin(\frac{\pi x_1}{a}) \sin(\frac{3\pi x_2}{a}) + \sin(\frac{3\pi x_1}{a}) \sin(\frac{\pi x_2}{a})], E_{13} = 10K \text{ (nondegenerate)}$$

Identical Fermions: $\psi_{13} = \frac{\sqrt{2}}{a} [\sin(\frac{\pi x_1}{a}) \sin(\frac{3\pi x_2}{a}) - \sin(\frac{3\pi x_1}{a}) \sin(\frac{\pi x_2}{a})]$, $E_{13} = 10K$ (nondegenerate)

$$\psi_{23} = \frac{\sqrt{2}}{a} [\sin(\frac{2\pi x_1}{a}) \sin(\frac{3\pi x_2}{a}) - \sin(\frac{3\pi x_1}{a}) \sin(\frac{2\pi x_2}{a})], E_{23} = 13K \text{ (nondegenerate)}$$

PROBLEM 5.5 (a) [5.19], with $\langle x^2 \rangle_n = \frac{a^2}{2}$ (Problem 2.5) and $\langle x^2 \rangle_n = a^2 (\frac{1}{3} - \frac{1}{2(n\pi)^2})$ (Problem 2.5) \Rightarrow

$$\langle (x_1 - x_2)^2 \rangle = a^2 (\frac{1}{3} - \frac{1}{2n^2\pi^2}) + a^2 (\frac{1}{3} - \frac{1}{2m^2\pi^2}) - 2 \frac{a^2}{2} \cdot \frac{a}{2} = \boxed{a^2 \left\{ \frac{1}{6} - \frac{1}{2\pi^2} \left(\frac{1}{n^2} + \frac{1}{m^2} \right) \right\}}$$