

Quiz 9

Sec 0101 $e^+e^- \rightarrow \gamma$



Momentum conservation says that the two photons leave back to back with equal magnitudes $p_{\gamma 1} = p_{\gamma 2}$. Energy conservation says $2m_e c^2 = E_{\gamma 1} + E_{\gamma 2}$. Since $E^2 = p^2 c^2 + (m_e c^2)^2$ and photons have no mass, $E_{\gamma} = p_{\gamma} c$. So $2m_e c^2 = 2E_{\gamma} \Rightarrow E_{\gamma} = m_e c^2$.

Sec 0104 ${}^2\text{H} + {}^2\text{H} \rightarrow {}^4\text{He}$

The hydrogens start at rest, so they both have only rest energy $E_H = m_H c^2$. The helium, by momentum conservation, must also be at rest, so it has energy $E_{He} = m_{He} c^2$. Energy conservation then says

$$2E_H = 2m_H c^2 = m_{He} c^2 + E_{out} = E_{He} + E_{out}$$

$$\text{So } E_{out} = (2m_H - m_{He})c^2$$

Sec 0105 $pp \rightarrow pp\pi^0$

For the smallest possible kinetic energy of the incoming protons, assume all outgoing particles are at rest.

Then energy conservation says

$$E_{in} + 2m_p c^2 = 2m_p c^2 + m_{\pi} c^2$$

$$\Rightarrow E_{in} = m_{\pi} c^2 \quad \text{For equal } K_{initial} \Rightarrow K_{final} = \frac{m_{\pi} c^2}{2}$$