Physics 132 Prof. W. Losert

Outline

Models of Light 1.Photons 2.Rays

Foothold Ideas: Light interacting with Matter

- Atoms and molecules naturally exist in states having specified energies. EM radiation can be absorbed or emitted by these atoms and molecules.
- When light interacts with matter, both energy and momentum are conserved.
- The energy of radiation either emitted or absorbed therefore corresponds to the <u>difference</u> of the energies of states.

Line Spectra





A molecule has the energy levels shown in the diagram at the right. We begin with a large number of these molecules in their ground states. We want to raise a lot of these molecules to the state labeled E_2 by shining light on it. What energy photon should we use?

- 1. 0.7 eV
- 2. 1.1 eV
- 3. **1.4 eV**
- 4. 1.8 eV
- 5. **2.1 eV**
- 6. **3.2** eV
- 7. Something else



A molecule has the energy levels shown in the diagram at the right. We have a large number of these molecules in the state E_2 . The state decays by emitting photons. What might we expect about the wavelength of the emitted photons?

- 1. They will be the same as the wavelength of the photons that were used to pump the molecules up to state E_2 .
- 2. Some might be the same wavelength, but some might be shorter.
- 3. Some might be the same wavelength, but some might be longer.
- 4. You only expect to see shorter wavelengths
- 5. You only expect to see longer wavelengths.
- 6. You will see longer, shorter, and the same wavelengths.

 $E_2 \xrightarrow{3.2 \text{ eV}}$ $E_1 \xrightarrow{1.8 \text{ eV}}$ $E_0 \xrightarrow{1.1 \text{ eV}}$

A molecule has the energy levels shown in the diagram at the right. We have a large number of these molecules in the state E_2 . The state decays by emitting photons. What energy photons might we expect to see?

1. 0.7 eV	1.B D F	F	3.2 eV
2. 1.1 eV	2.B D	E_2 -	
3. 1.4 eV	3.C		
1 1 8 0\/	4.C E	E_1 -	1.8 eV
-2.1 - 1/	5.ACE	Г	1.1 eV
5. 2.1 ev	6. Some other	E_0 -	
6. 3.2 eV	set		

In the transitions you found in the last slide, which corresponds to the longest wavelength? (and what is it)

1. 0.7 eV 2. 1.4 eV 3. 2.1 eV $f \lambda = c$ hc = 1234 eV-nm c = 3 10^8 m/s



- Foothold Ideas 1: Light as Rays - The Physics
- Through empty space (or ~air)
 light travels in straight lines.
- Each point on an object scatters light, spraying it off in all directions.
- A polished surface reflects rays back again according to the rule: The angle of incidence equals the angle of reflection.

Foothold Ideas 2: Light as Rays - the perception



 Our eyes identify a point as being on an object when rays traced back converge at that point.



Suppose you have a small brightly lit bulb, a mask (a cardboard screen with a small circular hole cut in it), and a screen. You see a small circle of light on the screen. What would happen to the spot if you moved the bulb straight upward a bit?

- 1. The spot would stay where it was.
- 2. The spot would move up a bit.
- 3. The spot would move down a bit.
- 4. The spot would move left a bit.
- 5. The spot would move right a bit.
- 6. Something else

PERSPECTIVE VIEW





Suppose you have two lit bulbs, the top one red and the bottom one blue, a mask (a cardboard screen with a small circular hole cut in it), and a screen, as shown. What would you see on the screen if you held the bulbs one over the other as shown?

- 1. One purple circle.
- 2. Two circles, one above the other with the top one red, the lower one blue.
- 3. Two circles, one above the other with the top one blue the lower one red.
- 4. Something else.





You are sitting in a chair looking at two objects that are suspended from the ceiling. It appears to you that object A is above object B. When you stand up, object A appears to be below object B. Which of the two objects is farther away from you?

- 1. Object A
- 2. Object B
- 3. They are both the same distance.
- 4. You can't tell. It could be either one



What you see

while sitting



B

A