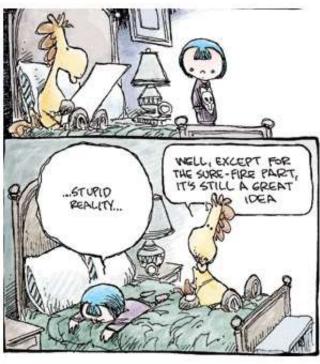
- Theme
 Music:
 Peggy Lee
 Fever
- Cartoon:
 Wiley
 NonSequitur





Definitions: Thermal energy

- Our model of matter as composed of many small moving particles allows us to extend energy conservation to include resistive forces.
- The energy associated with the motion of a single object is **coherent**; all parts of the object move in the same way.

 The object has a net momentum associated with its kinetic energy.
- The internal energy of an object is **incoherent**. The molecules of the object are moving in all directions randomly. Although the individual molecules have kinetic energy and momentum, the net momentum of the object as a result of its thermal energy is zero.

Definitions: Energy

- We can now expand our idea of energy to include more forms:
 - 1. Coherent energy of motion (kinetic) of the center of mass of an object: $\frac{1}{2} mv^2$
 - 2. Coherent energy of location relative to other objects (potential) of the center of mass.
 - 3. Incoherent internal energy of motion of the parts of an object (thermal)
 - 4. Submolecular energy of internal structure (chemical)

Definitions: Systems

■ If total energy of everything conserved, conservation isn't useful. What matters is how energy is moved around in relation to parts we care about.

■ Define systems:

- Isolated does not exchange energy or matter with the rest of the world.
- Closed exchanges energy but NOT matter with the rest of the world.
- Open exchanges both energy and matter with the rest of the world.

Exchange of Energy

- Energy can be transferred to or from a system (between "us" and "them") by
 - Work (coherent interaction via forces)
 - Heat (incoherent interaction via microscopic forces)

Equations

■ Total energy of a system (a set of macroscopic objects)

Internal energy

$$E = KE + PE + U$$

■ Exchanges of energy between the system and the rest of the universe

$$\Delta E = Q - W$$

Work done by the system on "them"

■ Exchanges of energy between the system and the rest of the universe ignoring coherent mechanical energy

$$\Delta U = Q - W$$

How to keep it straight

- This can be very confusing!

 There are many possible circumstances and the first law changes its form as a result.
- Keep a focus on the micro model, where energies can lie, and how they can be exchanged according to the laws of physics can help you keep things straight.