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Solutions by QJ

Hw Set 12: Ch 15: Q 11, 31, 53; Ex 11, 17, 23;
Ch 23: Q 4, 9; Ex 1, 6.

15: Q11. It takes 60 seconds for the seconds hand to go around once it has a time period of 60s. Its frequency = $\frac{1}{\text{period}} = \frac{1}{60} \text{ Hz} = 0.017 \text{ Hz}$.

15: Q31



15: Q53. There would be a maximum or an anti node at the mid point, because two maxima, or two minima, arrive there at the same time.


15: Ex 11. $T = 2\pi \sqrt{\frac{L}{g}} = 2\pi \sqrt{\frac{5 \text{ m}}{10 \text{ m/s}^2}} = 4.47 \text{ sec}$

15: Ex 17. $v = \lambda f = (25 \text{ cm}) (3 \text{ Hz}) = (0.25 \text{ m}) (3 \text{ Hz}) = 0.75 \text{ m/s} = v$

[NOTE] Ex. 22 was assigned in place of 23, by error, in F04.
15: Ex 23. The standing-wavelengths correspond to an integer no. of half wave lengths between the posts; therefore
 $\frac{\lambda_1}{2} = 3 \text{ m} \Rightarrow \lambda_1 = 6 \text{ m}$; $\frac{2\lambda_2}{2} = 3 \Rightarrow \lambda_2 = 3 \text{ m}$; $\lambda_3 = \frac{6}{3} = 2 \text{ m}$; $\lambda_4 = \frac{6}{4} = 1.5 \text{ m}$.
and $\frac{5\lambda_5}{2} = 3 \Rightarrow \lambda_5 = \frac{6}{5} = 1.2 \text{ m}$.
15: Ex 22. $f \cdot \lambda = v \Rightarrow f = \frac{2 \text{ m/sec}}{0.8 \text{ m}} = \frac{1}{T} = 2.5/\text{sec} \Rightarrow T = \frac{1 \text{ sec}}{2.5} = 0.4 \text{ sec}$

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Ch 23: Q 4, 9; Ex 1, 6

23 Q 4. Since Cl is in Col 7 of the periodic table (inside that of front row), its chemistry will resemble F, Br, I, and At, all of which lie in Column 7.

23 Q 9. Sample (a) contains elements A & B, but NOT C.
No, because every line in (a) occurs in either A or B

23: Ex 1 Cathode rays consist of free electrons. Therefore
$$\frac{\text{charge}}{\text{mass}} = q/m = -e/m_e = \frac{-1.6 \times 10^{-19} \text{ C}}{9.11 \times 10^{-31} \text{ kg}} = -1.76 \times 10^{11} \frac{\text{C}}{\text{kg}}$$

23: Ex 6 Nucleus is about 10^{-15} m across $\approx r_N$
Atom is about 10^{-10} m " $\approx r_A$. $\Rightarrow \frac{r_A}{r_N} = 10^5 = 100,000$ [cf p 490 line 1.]
Then if r_A is scaled up to 10 cm, r_N would be scaled up to $R = 10 \text{ cm} \times 10^{-5} = 10^{-4} \text{ cm} = 10^{-6} \text{ m} = 10^{-4} \mu\text{m} = 10^{-4} \text{ nm}$
As an order of magnitude estimate, this is in agreement with any value in the range $3 \text{ km} < R < 30 \text{ km}$.

Note added 12/7/05:

The above solution is based on the problem posed in the 4th edition of our text where the nucleus was represented by a ball of 10 cm radius. In the 5th edition the nucleus is modeled by a ball which is 1 cm across, 20x smaller. Then our model nucleus has a radius of 0.5 cm, so that our model atom has a radius $10^5 \times$ larger: $R = 0.5 \times 10^5 \text{ cm} = 5000 \text{ m} = 5 \text{ km}$. (Note that the scaling ratio $r_A/r_N = 10^5$ remains the same.)

end of HW # 12