

**Physics 115 – Section 0101**

**Homework #8**

**Due April 8, 2008**

Essay 1, 10 points

Suppose you are given a block of solid wax which is at room temperature. Describe how you would do an experiment to measure the specific heat of the wax, using things that might be in the 115 laboratory.

Essay 2, 10 points

Suppose the block of wax in Essay 1 melts at  $45^{\circ}\text{C}$ . Describe how you would do an experiment to measure the latent heat of the wax melting, using things that might be in the 115 laboratory.

Problem 1, 10 points

You mix 100g of ice at  $0^{\circ}\text{C}$  and 100g of hot water at  $55^{\circ}\text{C}$ . Describe the resultant mixture after it comes to equilibrium: What is the temperature? How much ice is there?

Problem 2, 10 points

How much hot water at  $55^{\circ}\text{C}$  do you need to add to 100g of ice at  $0^{\circ}\text{C}$  to end up with water at  $20^{\circ}\text{C}$ ?

Problem 3 10 points

How much steam at  $110^{\circ}\text{C}$  do you need to add to 50g of ice at  $-10^{\circ}\text{C}$  to end up with an ice water mixture containing 10g of ice? What would your answer be if you end up with water at  $20^{\circ}\text{C}$ ?

Essay 3

Explain why steam burns are more dangerous than water burns?